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Key indicators

Single-crystal X-ray study T = 292 K Mean σ (C–C) = 0.006 Å R factor = 0.052 wR factor = 0.113 Data-to-parameter ratio = 14.9

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

A new dinuclear Zn^{II} compound with a 1,4-benzenedicarboxylate bridge

In the compound μ_2 -1,4-benzenedicarboxylate- $\kappa^2 O, O'$ -bis-{aquachloro[4-(1*H*-1,3,7,8-tetraazacyclopenta[*I*]phenanthren-2-yl)phenol- $\kappa^2 N, N'$]zinc(II)} dihydrate, [Zn₂Cl₂(1,4-BDC)-(*L*)₂(H₂O)₂]·2H₂O, where 1,4-BDC is the 1,4-benzenedicarboxylate dianion, C₈H₄O₄²⁻, and *L* is 4-(1*H*-1,3,7,8tetraazacyclopenta[*I*]phenanthren-2-yl)phenol, C₁₉H₁₂N₄O, the Zn^{II} atom is five-coordinated by two N atoms from one *L* ligand, one Cl⁻ ion, an O atom from a 1,4-BDC anion and one water molecule in an approximate trigonal-bipyramidal geometry. The complete 1,4-BDC ligand is generated by inversion symmetry; it bridges two Zn^{II} atoms to form a dinuclear complex. Neighbouring complexes interact through π - π contacts and various kinds of hydrogen bonds, leading to a two-dimensional supramolecular structure.

Comment

The synthesis and design of supramolecular compounds possessing novel structures and properties have attracted increasing interest (Eddaoudi *et al.*, 2001). 1,10-Phenanthroline (phen) or its derivatives have long been used in the construction of metal–organic complexes. The phen derivative 4-(1H-1,3,7,8-tetraazacyclopenta[l]phenanthren-2-yl)phenol(*L*) possesses an extended delocalized aromatic system. To date, only a handful of supramolecular architectures based on *L* molecules have been described (Che *et al.*, 2006). We selected 1,4-benzenedicarboxylic acid (1,4-H₂BDC) as a linker and *L* as a secondary ligand, generating a new coordination compound, $[Zn_2(1,4-BDC)(L)_2(Cl_2)(H_2O)_2]\cdot 2H_2O$, (I), which is reported here.



In compound (I), the Zn^{II} atom is five-coordinated by two N atoms from one *L* ligand, one Cl⁻ ion, an O atom from a 1,4-BDC dianion and one water molecule (Fig. 1). Two Zn^{II} atoms are bridged by the centrosymmetric 1,4-BDC ligand to form a dinuclear complex. The Zn–N, Zn–Cl and Zn–O bond lengths are normal (Table 1). The ZnN₂O₂Cl coordination polyhedron approximates a trigonal bipyramid, with atoms N1 and O1 in the axial positions. The carboxylate C–O bond lengths suggest that the bonding is largely localized.

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metal-organic papers

It is notable that neighbouring dinuclear complexes interact through π - π contacts, leading to a two-dimensional supramolecular structure (Fig. 2). There are two types of π - π interactions: one exists between two *L* ligands (centroid separation = 3.45 Å), while the other exists between the *L* and 1,4-BDC ligands (centroid separation = 3.61 Å). Finally, various hydrogen bonds involving the water molecules, -NHgroups, Cl⁻ ions and carboxylate O atoms (Table 2) complete the crystal structure of (I).

Experimental

The ligand L was synthesized by the literature method of Steck & Day (1943). A methanol solution (6 ml) of L (0.5 mmol) was added slowly to an aqueous solution (12 ml) of $ZnCl_2 \cdot H_2O$ (0.5 mmol) and 1,4-H₂BDC (1 mmol) with stirring. The resulting solution was filtered and the filtrate was allowed to stand in air at room temperature for several days, yielding pale-yellow crystals of (I) in 61% yield based on Zn.

Crystal data [Zn₂Cl₂(C₈H₄O₄)(C₁₉H₁₂N₄O)₂- $\gamma = 104.02 \ (3)^{\circ}$ $(H_2O)_2]\cdot 2H_2O$ V = 1055.1 (5) Å³ Z = 1 $M_{\star} = 1062.47$ Triclinic, $P\overline{1}$ $D_r = 1.672 \text{ Mg m}^{-3}$ a = 9.1608 (18) Å Mo $K\alpha$ radiation b = 9.4769 (19) Å $\mu = 1.34 \text{ mm}^{-1}$ T = 292 (2) K c = 13.334 (3) Å $\alpha = 109.24 (3)^{\circ}$ Block, yellow $0.29 \times 0.25 \times 0.21 \ \text{mm}$ $\beta = 90.66 (3)^{\circ}$ Data collection Rigaku R-AXIS RAPID 10429 measured reflections

4780 independent reflections 3065 reflections with $I > 2\sigma(I)$

 $R_{\rm int}=0.059$

 $\theta_{\rm max} = 27.5^{\circ}$

 \overline{d} iffractometer ω scans Absorption correction: multi-scan (ABSCOR; Higashi, 1995) $T_{min} = 0.684, T_{max} = 0.758$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0392P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.052$	+ 0.7847P]
$wR(F^2) = 0.113$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.03	$(\Delta/\sigma)_{\rm max} = 0.001$
4780 reflections	$\Delta \rho_{\rm max} = 0.74 \text{ e } \text{\AA}^{-3}$
321 parameters	$\Delta \rho_{\rm min} = -0.43 \text{ e } \text{\AA}^{-3}$
H atoms treated by a mixture of	
independent and constrained	
refinement	

Table 1

Selected geometric parameters (Å, °).

Zn1-O1	2.049 (3)	Zn1-Cl1	2.2885 (14)
Zn1-O1W	2.080 (3)	C20-O2	1.237 (4)
Zn1-N1	2.150 (3)	C20-O1	1.263 (4)
Zn1-N2	2.072 (3)		
O1-Zn1-N2	90.17 (11)	O1W-Zn1-N1	89.60 (13)
O1-Zn1-O1W	87.43 (13)	O1-Zn1-Cl1	98.97 (10)
N2-Zn1-O1W	124.01 (13)	N2-Zn1-Cl1	118.99 (9)
O1-Zn1-N1	163.57 (12)	O1W-Zn1-Cl1	116.62 (11)
N2-Zn1-N1	78.13 (12)	N1-Zn1-Cl1	96.81 (9)



Figure 1

A view of (I). Displacement ellipsoids are drawn at the 50% probability level (arbitrary spheres for H atoms). [Symmetry code: (i) 1 - x, -y, 1 - z.]



Figure 2

A view of the two-dimensional supramolecular structure of (I) formed *via* π - π interactions. H atoms have been omitted.

Table 2 Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$\begin{array}{c} 01W - HW12 \cdots O2 \\ 01W - HW11 \cdots O2^{i} \\ 03 - H3A \cdots O2W^{ii} \\ N4 - H4 \cdots C11^{iii} \\ 02W - HW21 \cdots N3 \\ 02W - HW22 \cdots C11^{iv} \end{array}$	$\begin{array}{c} 0.87 (5) \\ 0.83 (6) \\ 0.82 \\ 0.86 \\ 0.88 (4) \\ 0.94 (2) \end{array}$	1.95 (6) 1.88 (6) 1.90 2.48 2.01 (2) 2.85 (4)	2.774 (5) 2.700 (5) 2.713 (5) 3.318 (4) 2.884 (5) 3.571 (4)	158 (4) 172 (6) 173 165 172 (6) 134 (5)

Symmetry codes: (i) -x + 2, -y + 1, -z + 1; (ii) -x, -y + 2, -z; (iii) x, y + 1, z; (iv) -x + 1, -y + 1, -z.

The H atoms of the water molecules were located in difference maps and their positions were freely refined; O–H distances are included in Table 2. For HW11 and HW12, U_{iso} was freely refined; for HW21 and HW22, the constraint $U_{iso}(H) = 1.5U_{eq}(O2W)$ was applied. The alcohol OH and all C– and N-bound H atoms were positioned geometrically (O–H = 0.82 Å, N–H = 0.86 Å and C–H = 0.93 Å) and refined as riding, with $U_{iso}(H) = 1.2U_{eq}(carrier)$.

Data collection: *PROCESS-AUTO* (Rigaku, 1998); cell refinement: *PROCESS-AUTO*; data reduction: *PROCESS-AUTO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL-Plus* (Sheldrick, 1990); software used to prepare material for publication: *SHELXL97*. The author thanks Jilin Normal University for supporting this work.

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